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APPLICATION NO. FILING DA		ILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.		
09/988,298		11/19/2001	William L. Bowden	08935-251001 / M-4971-Lam			
26171	7590	09/29/2003			•		
FISH & RICHARDSON P.C. 1425 K STREET, N.W. 11TH FLOOR WASHINGTON, DC 20005-3500				EXAM	EXAMINER		
				. WEINER, I	LAURA S		
				ART UNIT	PAPER NUMBER		
			•	1745			
				DATE MAILED: 09/29/2003			

Please find below and/or attached an Office communication concerning this application or proceeding.

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		Application No	D	Applicant(s)	7
	000	09/988,298		BOWDEN ET AL.	
	Office Action Summary	Examiner		Art Unit	
		Laura S Weiner		1745	
Period	Th MAILING DATE of this communication I for Reply	n appears on the cov	er sheet with the c	orrespondenc add	iress
T - - 8 - 1 - 1 - 6 - 6	SHORTENED STATUTORY PERIOD FOR R IE MAILING DATE OF THIS COMMUNICATION (Section 2) THIS COMMUNICATION (Section 2) THIS COMMUNICATION (Section 3) THIS COMMU	ON. FR 1.136(a). In no event, howers, howers, howers, howers, a reply within the statutory moveriod will apply and will expirestatute, cause the application	wever, may a reply be tim ninimum of thirty (30) days e SIX (6) MONTHS from to become ABANDONE	nely filed s will be considered timely the mailing date of this co O (35 U.S.C. § 133).	mmunication.
1)[	Responsive to communication(s) filed on	<u>19 November 2001</u>			
2a)[	☐ This action is <b>FINAL</b> . 2b)⊠	This action is non-	final.		
3)[ Dispo	<ul> <li>Since this application is in condition for a closed in accordance with the practice un sition of Claims</li> </ul>				e merits is
4)[	$\boxtimes$ Claim(s) <u>25</u> is/are pending in the applicat	ion.			
	4a) Of the above claim(s) 11-25 is/are with	ndrawn from conside	ration.		
5)[	Claim(s) is/are allowed.				
6)[	☑ Claim(s) <u>1-10</u> is/are rejected.				
7)[	Claim(s) is/are objected to.				
8)[	Claim(s) are subject to restriction a	and/or election requir	ement.		
Applic	cation Papers				
9)[	The specification is objected to by the Exa	miner.			
10)[	☐ The drawing(s) filed on is/are: a)☐	accepted or b)☐ obje	cted to by the Exar	miner.	
_	Applicant may not request that any objection				
11)[	The proposed drawing correction filed on _		, , , , , ,	ved by the Examine	er.
. 0. 6	If approved, corrected drawings are required		ction.		
, -	☐ The oath or declaration is objected to by th	ie Examiner.			
	ty under 35 U.S.C. §§ 119 and 120				•
13)[	Acknowledgment is made of a claim for for	oreign priority under (	35 U.S.C. § 119(a	)-(d) or (f).	
	a) All b) Some * c) None of:		·		
	1. Certified copies of the priority docur	ments have been red	eived.		
	2. Certified copies of the priority docur	ments have been red	eived in Application	on No	,
	<ul> <li>3. Copies of the certified copies of the application from the Internationa</li> <li>* See the attached detailed Office action for a</li> </ul>	al Bureau (PCT Rule	17.2(a)).		Stage
14)[	Acknowledgment is made of a claim for dor	nestic priority under	35 U.S.C. § 119(€	e) (to a provisional	application).
15)[	a) ☐ The translation of the foreign languag☐ Acknowledgment is made of a claim for do				
Attachn	_		30 ==		
2) 🔲 N	lotice of References Cited (PTO-892) lotice of Draftsperson's Patent Drawing Review (PTO-940 nformation Disclosure Statement(s) (PTO-1449) Paper No			r (PTO-413) Paper No(s Patent Application (PTC	

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### **DETAILED ACTION**

#### Election/Restrictions

1. Restriction to one of the following inventions is required under 35 U.S.C. 121:

- Claims 1-10, drawn to a primary lithium electrochemical cell, classified in class 429, subclass 224.
- II. Claims 11-23, 24-25, drawn to a method of preparing lambda-manganese dioxide and a method of manufacturing an electrochemical cell, classified in class 29, subclass 623.1.

The inventions are distinct, each from the other because of the following reasons:

- 2. Inventions I and II are related as process of making and product made. The inventions are distinct if either or both of the following can be shown: (1) that the process as claimed can be used to make other and materially different product or (2) that the product as claimed can be made by another and materially different process (MPEP § 806.05(f)). In the instant case, the product as claimed can be made by another and materially different process such as reacting Mn2O4 with an acid.
- 3. Because these inventions are distinct for the reasons given above and have acquired a separate status in the art as shown by their different classification, restriction for examination purposes as indicated is proper.

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4. During a telephone conversation with Mr. Fox on August 7, 2003, a provisional election was made with traverse to prosecute the invention of Group I, claims 1-10. Affirmation of this election must be made by applicant in replying to this Office action. Claims 11-25 are withdrawn from further consideration by the examiner, 37 CFR 1.142(b), as being drawn to a non-elected invention.

5. Applicant is reminded that upon the cancellation of claims to a non-elected invention, the inventorship must be amended in compliance with 37 CFR 1.48(b) if one or more of the currently named inventors is no longer an inventor of at least one claim remaining in the application. Any amendment of inventorship must be accompanied by a request under 37 CFR 1.48(b) and by the fee required under 37 CFR 1.17(i).

## Claim Rejections - 35 USC § 102

6. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

- (a) the invention was known or used by others in this country, or patented or described in a printed publication in this or a foreign country, before the invention thereof by the applicant for a patent.
- (b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.
- 7. Claims 1-5 are rejected under 35 U.S.C. 102(a) as being anticipated by Read et al. "Low Temperature Performance of Lambda-Manganese Oxide in Lithium Primary Batteries".

Read et al. teaches a lithium primary battery comprising a lambda-manganese dioxide cathode and a lithium anode having a higher energy density than conventional heat-treated B/gamma-MnO2 in primary lithium batteries over the temperature range of –40 degrees C to 40 degrees C and discharge rates from 0.1 to 2.0 mA/cm2. The improvement resulted from the increased voltage and improved discharge kinetics on the 4V plateau of lambda-MnO2.

## Claim Rejections - 35 USC § 103

- 8. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 9. Claims 1-3 are rejected under 35 U.S.C. 102(b) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Hunter (4,312,930).

Hunter teaches in column 6, lines 34-46, an electrochemical cell comprising a lithium anode, a cathode comprising lambda-manganese dioxide and an electrolyte. Hunter teaches in Figure 1, that the cell voltage starts at 4 V.

Since Hunter teaches the same lithium battery comprising a lambda-manganese dioxide positive electrode, a negative lithium electrode and an electrolyte then inherently the same lithium battery which contains an average closed circuit voltage of between

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3.8 an 4.1V and a specific discharge capacity to a 3V cutoff of greater than 130 mAh/g at a nominal discharge rate of 1 mA/cm2 must also be obtained.

In addition, the presently claimed property of battery which contains an average closed circuit voltage of between 3.8 an 4.1V and a specific discharge capacity to a 3V cutoff of greater than 130 mAh/g at a nominal discharge rate of 1 mA/cm2 would have obviously have been present once the Hunter product is provided. *In re Best, 195 USPQ 433 (CCPA 1977).* 

10. Claims 1-3 are rejected under 35 U.S.C. 102(b) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Giwa et al. "Lithium Primary Envelope Cells".

Giwa et al. teaches a primary battery comprising a lambda-manganese dioxide cathode and a lithium anode. Giwa et al. teaches in the conclusion section that lambda-manganese dioxide gives higher energy than standard MnO2 as half its discharge occurs on a higher voltage plateau (3.9V) while the remainder of the discharge is around 2.8 V.

Since Giwa et al. teaches the same lithium primary battery comprising a lambda-manganese dioxide positive electrode and a negative lithium electrode then inherently the same lithium battery which contains a specific discharge capacity to a 3V cutoff of greater than 130 mAh/g at a nominal discharge rate of 1 mA/cm2 must also be obtained.

In addition, the presently claimed property of battery which contains specific discharge capacity to a 3V cutoff of greater than 130 mAh/g at a nominal discharge rate

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of 1 mA/cm2 would have obviously have been present once the Giwa et al. product is provided. *In re Best, 195 USPQ 433 (CCPA 1977).* 

11. Claims 6-10 are rejected under 35 U.S.C. 103(a) as being unpatentable over Read et al. "Low Temperature Performance of Lambda-Manganese Oxide in Lithium Primary Batteries" in view of Furukawa et al. (5,294,499).

Read et al. teaches a lithium primary battery comprising a lambda-manganese dioxide cathode and a lithium anode having a higher energy density than conventional heat-treated B/gamma-MnO2 in primary lithium batteries over the temperature range of –40 degrees C to 40 degrees C and discharge rates from 0.1 to 2.0 mA/cm2. The improvement resulted from the increased voltage and improved discharge kinetics on the 4V plateau of lambda-MnO2.

Read et al. discloses the claimed invention except for specifically teaching that the lambda-MnO2 is maintained at a temperature of less than 150 degrees C during processing and that the lambda-MnO2 has a BET surface area of greater than 8 m2/g.

Furukawa et al. teaches in column 4, lines 26-50, that it is known to have a MnO2 having a BET of 41.6 m2/g and the grain sizes is of 0.1 to 20 um.

It would have been obvious to one having ordinary skill in the art at the time the invention was made to have a lambda-MnO2 having a BET surface area greater than 8 m2/g because Furukawa et al. teaches that this is known and since it has been held that

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where general conditions of a claim are disclosed in the prior art, discovering the optimum or workable ranges involves only routine skill in the art. *In re Aller, 105 USPQ* 233.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Laura S Weiner whose telephone number is 703-308-4396. The examiner can normally be reached on M-F (7:00-4:30).

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick Ryan can be reached on 703-308-2383. The fax phone number for the organization where this application or proceeding is assigned is (703) 872-9306.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is 703-308-0661.

Laura S Weiner Primary Examiner Art Unit 1745

September 8, 2003